

## Test Review

Draw lines to connect the matches.

- |                    |                        |
|--------------------|------------------------|
| a. $\text{g/cm}^3$ | 1. volume of a liquid  |
| b. mL              | 2. density of a liquid |
| c. $\text{cm}^3$   | 3. density of a solid  |
| d. g/mL            | 4. volume of a solid   |

1. What is volume?
2. What is density?
3. What is the volume of a box that is 14 cm long, 10 cm wide, and 3 cm high?
4. How do you find the volume of an irregularly shaped object?
5. Formula for Density:
6. If 96.5 grams of gold has a volume of 5 cubic centimeters, what is the density of gold?
7. If 96.5 grams of aluminum has a volume of  $35 \text{ cm}^3$ , what is the density of aluminum? How does its density compare with the density of gold?
8. If the density of a diamond is  $3.5 \text{ g / cm}^3$ , what would be the mass of a diamond whose volume is  $0.5 \text{ cm}^3$ ?